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~~Transport equation~~ Introduction to the Boltzmann transport equation (BTE) PGE 381M  
Lecture 2.1

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Momentum Transport lecture 3/10

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2\_8 LaPlace and Poisson Equations

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Statistical mechanics - Know It ALL ???

*Rarefied \u0026 Microscale Gases And Viscoelastic Fluids: A Unified Framework (Lecture- 1)* **Monte and Usha Ahuja**

**Distinguished Lecture Series on Feb. 14: Anil**

**K. Prinja** Francis Filbet: On hybrid method for rarefied gas dynamics: Boltzmann vs.

~~Navier-Stokes models~~ ~~Technical Session 5~~

~~Compressible Flows~~ ~~virtual room~~ Leone

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**ASEN 6061 Molecular Gas Dynamics and Direct  
MC Sim** Macroscopic Transport Equations For  
Rarefied

Thus, the proper simulation of flows in rarefied gases requires a more detailed description. This book discusses classical and modern methods to derive macroscopic transport equations for rarefied gases from the Boltzmann equation, for small and moderate Knudsen numbers, i.e. at and above the Navier-Stokes-Fourier level.

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Gas Flows Approximation Methods In Kinetic Theory  
Struchtrup H. (2005) Macroscopic transport equations for rarefied gas flows. In: Macroscopic Transport Equations for Rarefied Gas Flows. Interaction of Mechanics and Mathematics.

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MACROSCOPIC TRANSPORT MODELS FOR RAREFIED GAS FLOWS

7 of 26 The idea of the CE expansion method is to add corrections to the local equilibrium distribution by adding terms of higher orders in the Knudsen number,  $f = f^{(0)} + Kn f^{(1)} + Kn^2 f^{(2)} + \dots$ , (3.2) subject to the condition that the hydrodynamic variables  $\{\rho, v_i, \theta\}$  are the same at any level of expansion, so that  $\rho^{(1)}, v_i^{(1)}, \theta^{(1)} = m Z(1, c_i, C^2$

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Macroscopic transport models for rarefied gas flows: a ...

The main goal of this section is to study the nonlinear transport phenomena and macroscopic flow behavior of rarefied Couette flows from low speed to high speed, with particular concentration on the detailed structure of the nonisothermal KL and the shear-stress Knudsen number dependence of the effective transport coefficients in the whole system.

Nonlinear transport of rarefied Couette flows from low ...

Many macroscopic equations are proposed to

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describe the rarefied gas dynamics beyond the Navier-Stokes level, either from the mesoscopic Boltzmann equation or some physical arguments, including (i) Burnett, Woods, super-Burnett, augmented Burnett equations derived from the Chapman-Enskog expansion of the Boltzmann equation, (ii) Grad 13, regularized 13/26 moment equations, rational extended thermodynamics equations, and generalized hydrodynamic equations, where the velocity distribution ...

On the accuracy of macroscopic equations for linearized ...

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Struchtrup, H. 2005b Macroscopic Transport Equations for Rarefied Gas Flows. Springer.  
Struchtrup, H. 2012 Unique moment set from the order of magnitude method .

Macroscopic and kinetic modelling of rarefied polyatomic ...

The R13 equations, derived from the Boltzmann equation using the moment method, provide closure to the mass, momentum and energy conservation laws in the form of constitutive, transport equations for the stress and heat flux that extend the Navier-Stokes-Fourier model to include non-

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equilibrium effects. Methods In Kinetic Theory

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Thermophoresis of a spherical particle:

modelling through ...

The basis of most of the approaches to modeling of rarefied gases is the Boltzmann equation. In the continuum limit, a set of macroscopic transport equations can be obtained from the Boltzmann equation, and the Chapman-Enskog method utilized for their closure.

Influence of angular momentum on transport coefficients in ...

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Macroscopic transport equations for rarefied gas flows ...

Due to the failure of the continuum hypothesis for higher Knudsen numbers, rarefied gases and microflows of gases are

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particularly difficult to model. Macroscopic transport equations compete with particle methods, such as the Direct Simulation Monte Carlo method (DSMC), to find accurate solutions in ...

Evaporation Boundary Conditions for the Linear R13 ...

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Gas Flows ...

The kinetic theory of gases is a historically  
significant, but simple, model of the  
thermodynamic behavior of gases, with which  
many principal concepts of thermodynamics  
were established. The model describes a gas as  
a large number of identical submicroscopic  
particles (atoms or molecules), all of which



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are in constant, rapid, random motion. Their size is assumed to be much smaller than the  
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